

***In-situ* investigation of deuterium sorption mechanisms in forged Mg-Mg₂Ni composites**

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In the search for new materials for hydrogen storage, we investigate low cost Mg-Mg₂Ni composite materials. Samples processed by annealing and fast forging show a particular microstructure and a remarkable improvement of the hydrogen sorption kinetics.

In situ neutron diffraction was performed on CRG- D1b (ILL, Grenoble) to investigate the deuterium sorption mechanisms in Mg-Mg₂Ni composite materials. Neutron powder diffraction patterns were collected upon successive absorptions and desorptions while monitoring the deuterium pressure.

The sequence of phase transition for the first deuteration is very different from the subsequent ones. Upon the first deuteration (initial activation), deuterium evolution proceeds through the instantaneous formation of the solid solution Mg₂NiD_{0.3} followed by almost simultaneously formation of the two deuteride phases MgD₂ and Mg₂NiD₄. The rate-limiting step for the first deuterium uptake is considered to be surface nucleation of the deuterides. Upon the second/third deuteration, a two-stage deuterium uptake accompanied by the sequential formation of deuterides occurred, where only the MgD₂ was formed during the first 17 min due to favorable thermodynamics and kinetics. Mg₂NiD₄ started to form once the formation of MgD₂ became gradually saturated.

We will show how *in situ* neutron powder diffraction coupled with sorption measurements and combined to kinetics analysis and microstructure observations uncover the sorption mechanisms in promising Mg-Mg₂Ni composite materials.